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top-2-top 100 book pdf ebook write a book a game made beowulf and heroquest pdf. Top 1000 Engineers. Top 100 Cool Tools. WOW What a video!. cool that you got the Iha vy in kindle format. Wow! What a video!. dogman how he do that. This invention relates to the formation of hydrogen fluoride from calcium fluoride and hydrofluoric acid. Hydrogen fluoride has a number of uses, among which may be mentioned the manufacture of fluorohydric acids and their partial oxidation to obtain compounds with fluorine content, for example fluoroethyl alcohols; as a stabiliser in rocket fuels, additives in paints and roofing, synthetic forage-fertilizer, etc. The synthesis of hydrogen fluoride (HF) from calcium

fluoride (CaF_2) and hydrofluoric acid is known in the art. This reaction is carried out in liquid or gaseous hydrofluoric acid. The prior art process comprising the reaction of CaF_2 with HF in liquid H_2O containing either Na_2CO_3 or Na_2SO_4 gives good yields. However, the dissolution of CaF_2 produces floccules of calcium hydroxide which precipitate rapidly giving, on dissolution, a pulp or sludge, containing flocs of calcium hydroxide and a mass of silica. It is therefore necessary to have a large bed of finely divided CaF_2 in order to obtain a reasonable yield of HF. The prior art processes also suffer from various disadvantages among which may be mentioned the formation of CaF_2 flocs which are difficult to break up and the generation of water-soluble H_3PO_4 in the process solution which in turn causes precipitation of basic salts in the process. A process for the synthesis of HF by the reaction of CaF_2 with HF in a gaseous atmosphere is disclosed in U.S. Pat. No. 3,132,037. The process comprises the reaction of CaF_2 and H_2F in a gaseous atmosphere at an elevated temperature. The gaseous H_2F is produced by passing HF in gaseous form into a closed vessel containing CaF_2 at a temperature of

from 140.degree